

# Scientific Measurement

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 The metric system is a <u>decimal</u> system of measurement whose units are based on multiples of <u>10</u>

The basic unit of length, the meter, can be multiplied or divided to measure objects and distances much larger or smaller than a meter.

The same process can be used when measuring volume and mass.

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Temperature (degrees <u>Celsius</u>) = The amount of <u>heat</u> in something

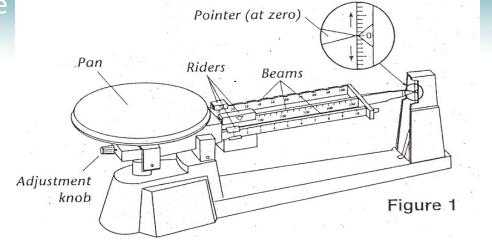
### **Finding Mass**

Mass can be found using either a digital scale or a triple-beam balance

When using a digital balance, be sure to zero the scale and then place the object on the pan



### **Finding Mass**



Triple-beam balances have three beams that show the mass of the object.

- The middle beam is 0-500 grams and the rider is 100 grams
- The back beam is 0-100 grams and the rider is 10 grams
- The front beam is 0-10 grams and the rider is 1 gram

To find the mass of an object you place the object on the pan and move the riders until the pointer on the right of the balance stays pointed to zero

> Be sure that the pointer is at zero before you place the object on the pan

### **Finding Mass**

#### SOLID...

- Place object on the pan and move the riders until the pointer stays at zero
- Add up the numbers on the beams where the riders are positioned to find the mass

#### LIQUID...

- Measure the mass of an empty container that can hold the liquid
- Measure the combined mass of the container and the liquid
- Subtract the mass of the container from the combined mass

#### CHEMICAL...

- Find the mass of a piece of weighing paper or empty container that will hold the chemical
- Add this amount to the desired mass of the chemical and preset the riders to this number
- Add the chemical to the weighing paper a little at a time until the pointer points to zero.

### Finding Volume

To find the volume of a regular object, such as a block, you multiply the length by the width by the height (Volume = LxWxH)

Cubic centimeter

To find the volume of a liquid you would use a graduated cylinder

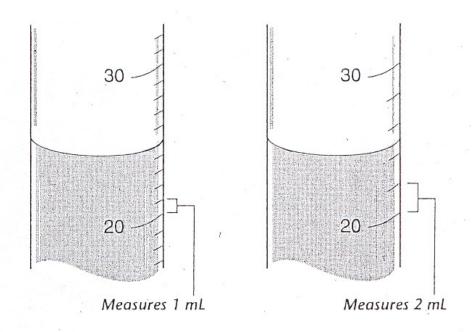
Liter, milliliter, etc.

Always read a graduated cylinder at eye level and from the bottom of the meniscus (the curved surface)

> 46 mL meniscus

### **Finding Volume**

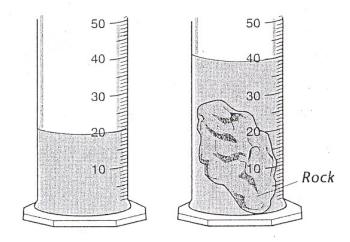
Always check the unnumbered marks on a graduated cylinder to see how many sections there are and what they measure. Also, sometimes you have to estimate a measurement between two marks. You should find that both graduated cylinders to the right contain 25mL.



### **Finding Volume**

To find the volume of an irregular object you would use a graduated cylinder

- Add enough water to the graduated cylinder to cover the object
- Hold the graduated cylinder at an angle (to prevent cracking and splashing) and place the object into the graduated cylinder
- Slowly bring the graduated cylinder upright
- The volume of the object is the difference in water levels before and after the object was added



### **Finding Density**

Density is the ratio of mass to volume

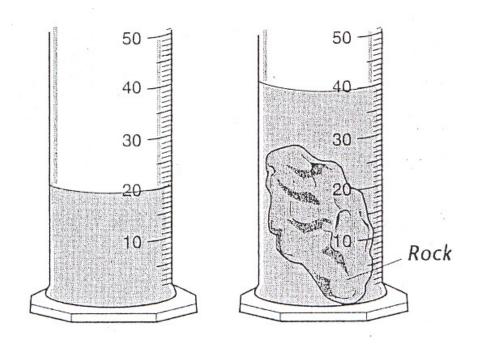
The density of a specific kind of matter helps to identify it and to distinguish it from other kinds of matter

The formula for density is mass divided by volume

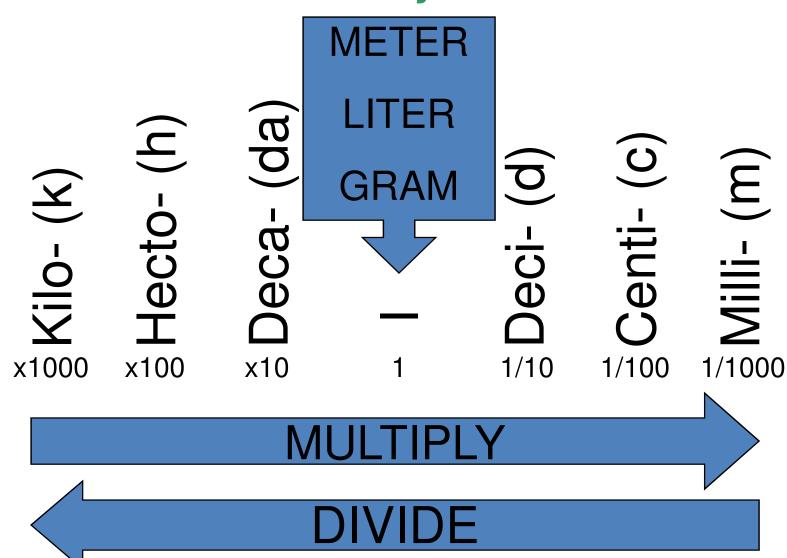
Density = Mass / Volume

### **Finding Density**

If the mass of the rock below is 40 grams, what is the density of the rock?



### **Prefixes in the Metric System**



A bar graph is used to display data in a number of separate, or distinct, <u>categories</u>

Example = Types of pets that students own (the categories would be dog, cat, snake, etc.)

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 Line graphs are useful because you can <u>estimate</u> values for conditions that you did not test in the experiment

Example = If you measured 1°C and 2°C, using a line graph you could predict 1.5°C